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Text Structure and Graphic Sources



Directions: Read the article and look at the picture. Then read each question about the article and picture. Choose the best answer to the question. Mark the space for the answer you have chosen.

Girl Saves Cat Trapped in Snow
 Caitlin Smith of Wildflower Road was walking home from school yesterday afternoon when she heard something crying. She thought something might be trapped in the snow.
 She found the spot where she heard the sound and began to dig in the snow. "I reached in and pulled out a cat!" Caitlin said. "The poor thing was wet and cold, and it wouldn't stop crying."
 Caitlin brought the cat to the vet. If no one claims the cat, Caitlin's parents say she can keep her.



Caitlin Smith dug this hole to free the trapped cat.

1. The article tells about—
- Wildflower Road.
 - the dangers of snow.
 - how to care for your pet.
 - a girl's rescue of a cat.

2. The author of this article—
- used Caitlin's words.
 - made up a story.
 - guessed what happened.
 - found the cat.
3. This story tells what happened—
- in a mixed-up order.
 - last week.
 - in the order that the events occurred.
 - in a backward order.
4. The picture shows—
- Caitlin's house.
 - Caitlin's school.
 - how the cat got trapped.
 - the hole that Caitlin dug.
5. This article probably appeared—
- in a local newspaper.
 - in a school textbook.
 - on a neighborhood sign.
 - in a storybook.

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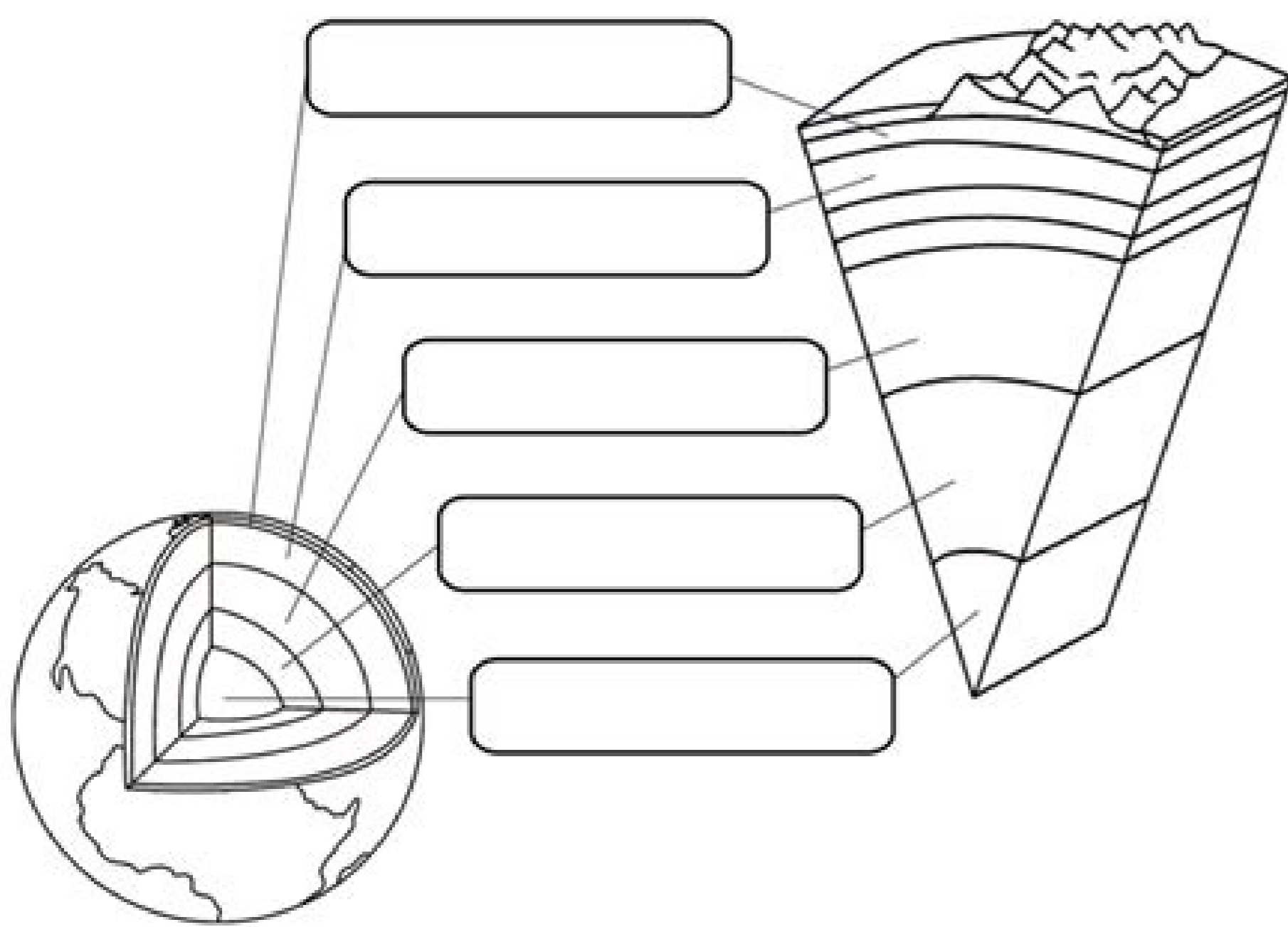


Notes for Home: Your child reviewed elements of text structure. **Home Activity:** Read a newspaper or magazine article with your child. Talk about how the author interviews real people and reports what they say in order to tell the story.

Name: _____

Date: _____

Identify the
Different Layers of the Earth



Word Box

Crust Upper mantle

Lower mantle Outer core Inner core

Atomic Structure Practice

Part I: Reading the Periodic Table

The following table includes information about the name, symbol, atomic number, atomic mass, and numbers of protons, neutrons, and electrons for several elements. Use your periodic table to fill in all of the blanks.

Element Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons	
Lithium	Li	3	6.941	3	4	3
Hydrogen		17				
	Au					
		30.974				
Tungsten				30		
		126.90				
	14					

Part II: Electron Configuration

Determine how many electrons are contained in the outer orbital of each atom below. (Use your periodic table to determine the total number of electrons)

- | | | |
|--------------------|-----------------|----------|
| 5 _____ Phosphorus | _____ Hydrogen | _____ Ne |
| _____ Lithium | _____ Chlorine | _____ Na |
| _____ Helium | _____ Magnesium | _____ F |
| _____ Oxygen | _____ Boron | _____ C |

Two of the above elements have stable electron configurations. Circle them.

Part III: Ions

Use your periodic table to determine the number of electrons and protons found in each of the following ions.

	Protons	Electrons		Protons	Electrons
Na ⁺	11	10	S ²⁻		
O ²⁻			H ⁺		
Li ⁺			Ca ²⁺		
Mg ²⁺			F		

Chapter 14
 Molecular and Ionic
 Equilibrium and Kinetics
 Total Review questions: 100
 1. _____ 2. _____ 3. _____

4. _____ 5. _____ 6. _____

7. _____ 8. _____ 9. _____

10. _____ 11. _____ 12. _____

13. _____ 14. _____ 15. _____

16. _____ 17. _____ 18. _____

19. _____ 20. _____ 21. _____

Total Review questions: 100
 22. _____ 23. _____

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